

## Experiment 4B: The rotational-vibrational spectrum of HCl

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### Abstract

Experiment abstract.

## 1 Results and Analysis

Table 1: Rotational absorbances for the fundamental transition.

	$J$	$R(J) / \text{cm}^{-1}$	$P(J) / \text{cm}^{-1}$	$R(J) - P(J) / \text{cm}^{-1}$	$R(J - 1) - P(J + 1) / \text{cm}^{-1}$
H <sup>35</sup> Cl	0	2906.4			
	1	2926.03	2865.24	60.79	62.64
	2	2945.05	2843.76	101.29	104.33
	3	2963.42	2821.70	141.72	145.98
	4	2981.14	2799.07	182.07	187.53
	5	2998.19	2775.89	222.30	228.97
	6	3014.57	2752.17	262.40	270.26
	7	3030.27	2727.93	302.34	311.4
	8	3045.28	2703.17	342.11	352.37
	9	3059.52	2677.90	381.62	393.13
	10	3073.10	2652.15	420.95	433.62
	11	3085.88	2625.90	459.98	473.88
	12		2599.22		
H <sup>37</sup> Cl	0	2904.29			
	1	2923.91	2863.20	60.71	62.53
	2	2942.90	2841.76	101.14	104.19
	3	2961.24	2819.72	141.52	145.76
	4	2978.93	2797.14	181.79	187.25
	5	2995.96	2773.99	221.97	228.62
	6	3012.32	2750.31	262.01	269.86
	7	3027.98	2726.10	301.88	310.95
	8	3042.97	2701.37	341.60	351.84
	9	3057.22	2676.14	381.08	392.57
	10	3070.73	2650.40	420.33	432.99
	11	3083.51	2624.23	459.28	473.20
	12		2597.53		

The values of  $\tilde{B}_0$ ,  $\tilde{B}_1$  and  $\tilde{B}_2$  were calculated accounting for the centrifugal distortion of the molecules by using equations 1.0.1 and 1.0.2.

$$\begin{aligned}
 R(J) - P(J) &= (4\tilde{B}_1 - 6\tilde{D}_1)(J + \frac{1}{2}) - 8\tilde{D}_1(J + \frac{1}{2})^3 \\
 \frac{R(J) - P(J)}{J + \frac{1}{2}} &= -8\tilde{D}_1(J + \frac{1}{2})^2 + 4\tilde{B}_1 - 6\tilde{D}_1
 \end{aligned}
 \tag{1.0.1}$$

A similar rearrangement for the transitions with a common excited vibrational state yields equation 1.0.2.

Table 2: Rotational absorbances for the overtone transition.

	$J$	$R(J) / \text{cm}^{-1}$	$P(J) / \text{cm}^{-1}$	$R(J) - P(J) / \text{cm}^{-1}$	$R(J - 1) - P(J + 1) / \text{cm}^{-1}$
$\text{H}^{35}\text{Cl}$	0	5688.16			
	1	5706.59	5647.71	58.88	62.66
	2	5723.78	5625.50	98.28	104.35
	3	5739.75	5602.24	137.51	145.98
	4	5754.53	5577.80	176.73	187.53
	5	5767.89	5552.22	215.67	229.05
	6	5780.07	5525.48	254.59	270.26
	7	5790.92	5497.63	293.29	311.52
	8		5468.55		
$\text{H}^{37}\text{Cl}$	0	5684.12			
	1	5703.13	5643.60	59.53	62.58
	2	5719.67	5621.54	98.13	104.80
	3	5735.61	5598.33	137.28	145.76
	4	5750.30	5573.91	176.39	197.19
	5	5763.27	5548.42	214.85	228.64
	6	5775.12	5521.66	253.46	269.43
	7	5786.65	5493.84	292.81	310.49
	8	5464.63			

$$\frac{R(J - 1) - P(J + 1)}{J + \frac{1}{2}} = -8\tilde{D}_0\left(J + \frac{1}{2}\right)^2 + 4\tilde{B}_0 - 6\tilde{D}_0 \quad (1.0.2)$$

Graphs of  $\frac{R(J)-P(J)}{J+\frac{1}{2}}$  and respectively  $\frac{R(J-1)-P(J+1)}{J+\frac{1}{2}}$  were plotted against  $(J + \frac{1}{2})^2$  and linear regressions were performed in order to determine the values for the centrifugal distortion coefficients,  $\tilde{D}_\nu$ , shown in table 4 and hence the rotational constants,  $\tilde{B}_\nu$ , shown in table 3.

The bond lengths were calculated from the  $\tilde{B}_\nu$  values using equation 1.0.3.

$$r_\nu = \sqrt{4\pi c \hbar \mu \tilde{B}_\nu} \quad (1.0.3)$$

Table 3: Rotational constants and bond lengths.

	$\nu$	$\tilde{B}_\nu / \text{cm}^{-1}$	$r_\nu / \text{pm}$
$^{35}\text{Cl}$	0		
	1		
	2		
$^{37}\text{Cl}$	0		
	1		
	2		

The error propagation shown in equation 1.0.4 was completed to estimate the uncertainty in the values of  $\tilde{B}_\nu$  where  $\alpha_B$ ,  $\alpha_D$ ,  $\alpha_m$  are the uncertainties in  $\tilde{B}_\nu$ ,  $\tilde{D}_\nu$  and the gradient found in the linear regression respectively.

$$\alpha_B = \tilde{B}_\nu \sqrt{\left(\frac{\alpha_m}{m}\right)^2 + \left(\frac{\alpha_D}{\tilde{D}_\nu}\right)^2} \quad (1.0.4)$$

Furthermore the calculus-based approximation<sup>1</sup> was utilised to give the estimation in the uncertainty for  $r_\nu$  ( $\alpha_r$ ) given in equation 1.0.5 since the  $\alpha_B$  values are small.

$$\alpha_r = \sqrt{\frac{\pi c \hbar \mu}{\tilde{B}_\nu}} \alpha_B \quad (1.0.5)$$

### 1.1 Determination of Vibrational Constants

The values for the harmonic constant  $\tilde{\nu}_e$  and the dimensionless anharmonicity constant  $x_e$  in equation 1.1.1 (which gives the expected vibrational energy levels if the rotational spectrum is ignored).

$$\tilde{E}_\nu = \tilde{\nu}_e \left( \nu + \frac{1}{2} \right) - \tilde{\nu}_e x_e \left( \nu + \frac{1}{2} \right)^2 \quad (1.1.1)$$

From equation 1.1.1 we can obtain the energy related to the pure vibrational transition  $\tilde{E}(\nu_f \leftarrow 0)$  (the  $\nu = 0$  to  $\nu = \nu_f$  transition) as equation 1.1.2.

$$\begin{aligned} \tilde{E}(\nu_f \leftarrow 0) &= \tilde{E}_{\nu_f} - \tilde{E}_0 \\ &= \tilde{\nu}_e \left( \nu_f + \frac{1}{2} - \frac{1}{2} \right) - \tilde{\nu}_e x_e \left( \frac{1}{4} - \left( \nu_f + \frac{1}{2} \right)^2 \right) \\ &= \tilde{\nu}_e (\nu_f - (\nu_f^2 + \nu_f) x_e) \end{aligned} \quad (1.1.2)$$

The energy related to the R branch transitions can be determined to yield equation 1.1.3 where  $J$  is the rotational state adopted in the lower vibrational state ( $\nu = 0$ ).

$$R(J) = \Delta \tilde{E}(\nu_f \leftarrow 0) + (\tilde{B}_{\nu_f} + \tilde{B}_0)(J+1) + (\tilde{B}_{\nu_f} - \tilde{B}_0)(J+1)^2 \quad (1.1.3)$$

Setting  $J = 0$  hence gives equation 1.1.4.

$$R(J) - 2\tilde{B}_{\nu_f} = \Delta \tilde{E}(\nu_f \leftarrow 0) \quad (1.1.4)$$

Substituting equation 1.1.2 into equation 1.1.4 yields equation 1.1.5.

$$R(0) - 2\tilde{B}_{\nu_f} = \tilde{\nu}_e (\nu_f - (\nu_f^2 + \nu_f) x_e) \quad (1.1.5)$$

Now from equation 1.1.5 a system of linear equations can be obtained by setting  $\nu_f = 1$  and  $\nu_f = 2$  (equations xx and xx) for which we can solve to give equations xx and xx.

for which setting  $J = 0$  and  $\nu_f = 1$  and subsequently  $\nu_f = 2$  gives the system of linear equations (equations xx and xx) for which we can solve to give equations xx and xx.

## 2 Discussion

Two methods able to be used to determine the value of  $\tilde{B}_0$ : using either the absorption data from the fundamental or the overtone transition. Much larger signal:noise ratio for overtone due to lower probability of transition, hence greater uncertainty in this value.

## References

- (1) I. G. Hughes and T. P. A. Hase, *Measurements and their Uncertainties*, Oxford University Press, Oxford, 2010.

## 3 Supplementary Information

Table 4: Centrifugal distortion coefficients.

	$\nu$	$\tilde{D}_\nu / 10^{-4} \text{ cm}^{-1}$
$^{35}\text{Cl}$	0	$-5.25 \pm 0.04$ and $-5.4 \pm 0.3$
	1	$-5.11 \pm 0.02$
	2	$-4.3 \pm 0.8$
$^{37}\text{Cl}$	0	$-5.20 \pm 0.03$ and $-9 \pm 2$
	1	$-5.13 \pm 0.01$
	2	$-11 \pm 4$